Final exam 8.2.2013

1. Are the following sentences true or false? If a sentence is false, explain shortly why.
a) The $r$ process proceeds close to stable nuclei. [2 p]
b) The vp process takes place in core-collapse supernovae. [2 p]
c) Uranium isotopes are produced via s process. [2 p]
d) ${ }^{12} \mathrm{C}$ is not consumed in the CNO cycle. [2 p]
2. a) Consider the direct capture reaction ${ }^{15} \mathrm{~N}(\mathrm{n}, \gamma)^{16} \mathrm{~N}$. Assume only direct capture into the ground (2) and the first three excited states at $120.42 \mathrm{keV}\left(0^{-}\right), 298.22 \mathrm{keV}\left(3^{-}\right)$ and $397.27 \mathrm{keV}\left(1^{-}\right)$in ${ }^{16} \mathrm{~N}$ plays a role. What is the dominant direct capture reaction mechanism in terms of orbital angular momentum of the neutron (s-wave, p-wave,...) and multipolarity of the emitted gamma ray (E1, M1, E2,...)? The groud-state spin of ${ }^{15} \mathrm{~N}$ is $1 / 2^{+}$. [5 p]
b) Above which minimum excitation energy would a state in ${ }^{16} \mathrm{~N}$ have to be located to serve as a resonance in the ${ }^{15} \mathrm{~N}(\mathrm{n}, \mathrm{\gamma})$ reaction? [3 p]
3. a) Explain the different steps of the ppl chain. [3 p]
b) What is the bottleneck reaction (the slowest reaction) in the ppl chain? [1 p]
c) How much energy is released in the ppl chain? You can use the attached mass tables to calculate that. [1 p]
d) Assuming that the ppl chain is totally responsible for the production of solar neutrinos, estimate the flux of solar neutrinos on Earth. The distance from Earth to Sun is about $1.5 \cdot 10^{8} \mathrm{~km}$.[3 p]
4. The following picture (Iliadis, Fig. 1.2.a) shows the solar abundance pattern as a

function of mass number A. Write a short essay to explain the main features of this distribution. [8 p]
5. a) Estimate a cross section for the reaction ${ }^{44} \mathrm{Ti}(\alpha, \mathrm{p})^{47} \mathrm{~V}$. You have detected 10 protons with your detector setup which has an efficiency of $20 \%$. A $2-\mathrm{cm}$-long helium gas target with a target thickness of about $20 \mu \mathrm{~g} / \mathrm{cm}^{2}$ has been bombarded by a ${ }^{44} \mathrm{Ti}^{13+}$ beam with an intensity of 6.0 pA (electrical current) for 8 hours. [ 6 p ]
b) Was the cross section for the reaction ${ }^{44} \mathrm{Ti}(\alpha, \mathrm{p})^{47} \mathrm{~V}$ discussed in a) measured in normal kinematics or in inverse kinematics? [1 p]
c) Why some reactions are measured in inverse kinematics instead of normal kinematics? [1 p]
6. a) About 3 s after the onset of the Big Bang, the neutron-proton ratio became frozen when the temperature was still as high as $10^{10} \mathrm{~K}$. About 250 s later, fusion reactions took place converting neutrons and protons into ${ }^{4} \mathrm{He}$. Essentially all neutrons were converted to ${ }^{4} \mathrm{He}$. Calculate the abundances of ${ }^{1} \mathrm{H}$ and ${ }^{4} \mathrm{He}$ after the primordial nucleosynthesis. The neutron half-life is 10.24 min and the neutron-proton mass difference is $1.29 \mathrm{MeV} / \mathrm{c}^{2}$. [4 p]
b) Xenon has nine stable isotopes between $A=124-136$. Which of these isotopes are produced by (i) p process, (ii) s process and (iii) r process? Note that some of the isotopes can be produced both via $s$ and $r$ process. Half-lives are given for the nuclei. "+" indicates $\beta^{+}$decay, otherwise $\beta$ - decay for the non-stable nuclei. [4 p]?

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| I | ${ }_{+}^{+122}$ | +123 13 h | +124 4 d | + 125 59 | + <br> 126 <br> 13 d | 127 | $\begin{aligned} & 128 \\ & 25 \end{aligned}$ | $\begin{aligned} & 129 \\ & 10^{7} a \end{aligned}$ | 130 12 h | 131 8 d | $\begin{aligned} & 132 \\ & 2 \mathrm{~h} \end{aligned}$ | $\begin{aligned} & 133 \\ & 21 \mathrm{~h} \end{aligned}$ | $\begin{gathered} \hline 134 \\ 52 \end{gathered}$ | $\begin{aligned} & 135 \\ & 7 \mathrm{~h} \end{aligned}$ | $\begin{aligned} & 136 \\ & 84 \mathrm{~s} \end{aligned}$ |
| Te | $\begin{array}{r} \hline 121 \\ \hline 17 d \\ \hline \end{array}$ | 122 | 123 | 124 | 125 | 126 | $\begin{gathered} 127 \\ 9 \mathrm{~h} \\ \hline \end{gathered}$ | 128 | 129 <br> 70 <br> min <br> 128 | 130 | $\begin{gathered} 131 \\ 25 \\ \text { min } \end{gathered}$ | $\begin{aligned} & 132 \\ & 76 \mathrm{~h} \\ & \hline \end{aligned}$ | $\begin{gathered} 133 \\ 13 \\ 13 \\ \text { min } \end{gathered}$ | 134 <br> 42 <br> min <br> 1 | $\begin{array}{r} 135 \\ 19 \mathrm{~s} \\ \hline \end{array}$ |
| Sb | +120 16 min | 121 | $\begin{array}{r} 122 \\ 3 \mathrm{~d} \\ \hline \end{array}$ | 123 | $\begin{aligned} & 124 \\ & 60 \mathrm{~d} \\ & \hline \end{aligned}$ | $\begin{array}{r} 125 \\ 3 \mathrm{a} \\ \hline \end{array}$ | $\begin{aligned} & 126 \\ & 12 \mathrm{~d} \end{aligned}$ | 127 4 d | $\begin{aligned} & 128 \\ & 9 \mathrm{~h} \\ & \hline \end{aligned}$ | $\begin{gathered} 129 \\ 4 \mathrm{~h} \\ \hline \end{gathered}$ | $\begin{aligned} & 130 \\ & 6 \text { min } \end{aligned}$ | 131 23 min | $\begin{aligned} & 132 \\ & 3 \text { min } \end{aligned}$ | $\begin{aligned} & 133 \\ & 3 \text { min } \end{aligned}$ | $\begin{aligned} & 134 \\ & 0.75 \mathrm{~s} \end{aligned}$ |

